New Eudesmane Sesquiterpenes from the Root of Lindera strychnifolia

Isao Kouno,* Asuka Hirai, Akiko Fukushige, Zhi-Hong Jiang, and Takashi Tanaka

Faculty of Pharmaceutical Sciences, Nagasaki University, Bunkyo-machi 1-14, Nagasaki 852-8521, Japan

Received March 29, 2000

Strychnistenolide (1) and its acetate 2 were isolated from the root of *Lindera strychnifolia*, along with a novel rearranged type of secoeudesmane, strychnilactone (3). Their structures were elucidated by extensive analysis of their NMR spectra, including 2D NMR techniques, together with an X-ray analysis for 3. Strychnistenolide exists as a single stereoisomer in $CHCl_3$, but in pyridine is epimerized.

The root of *Lindera strychnifolia* (Sieb. & Zucc.) f. Villars (Lauraceae) has a strong fragrance and is used in Chinese folk medicine as a palliative and an antispasmodic. In previous studies on the constituents from this plant, about 20 eudesmanes and elemanes were isolated and structurally elucidated by Takeda et al.^{1–3} Further studies on the chemical constituents of this species led to the isolation of bisesquiterpene⁴ and rearranged secoeudesmanolide.⁵ This paper describes the isolation and structure determination of two new eudesmane sesquiterpenes, **1** and **2**, and a rearranged secoeudesmanolide, **3**.

Results and Discussion

Compound **1** displayed a molecular ion peak at m/z 262 in MS and 15 carbon signals in its ¹³C NMR spectrum. These, together with elemental analysis, established the molecular formula of C₁₅H₁₈O₄.

The ¹H NMR spectrum in CDCl₃ (see Experimental Section) of compound 1 indicated the presence of a vinylidene group at $\delta_{\rm H}$ 5.14 and 5.11, an olefinic methyl group at $\delta_{\rm H}$ 1.92, a tertiary methyl group at $\delta_{\rm H}$ 0.52, and a methine group having an oxygen function at $\delta_{\rm H}$ 4.42. However, the ¹H and ¹³C NMR spectra including 2D NMR spectra, when taken in C₅D₅N (Table 1), displayed duplicate signals. These spectra revealed that in C₅D₅N solution 1 is an isomeric mixture of 1A and 1B in a ratio of approximately 6:7. Therefore, each of the signals of the two isomers was assigned individually by interpretation of the 2D NMR spectra (COSY, HSQC, and HMBC). The ¹H and ¹³C NMR spectra along with the interpretation of COSY, HSQC, and HMBC spectra indicated the presence of a cyclopropane ring moiety and a vinylidene group adjacent to the cyclopropane ring. The HMBC spectrum showed connectivities between the C-14 methyl carbon signal and H-1 proton signal of the cyclopropane ring.

The ¹³C NMR spectrum showed two lactone carbonyl carbon signals at $\delta_{\rm C}$ 171.7 (**1A**) and $\delta_{\rm C}$ 171.1 (**1B**) and two acetal carbon signals at $\delta_{\rm C}$ 105.7 (**1A**) and $\delta_{\rm C}$ 105.9 (**1B**). In the HMBC spectrum, the signals at $\delta_{\rm H}$ 4.82 (d, J = 10.5 Hz, H-6, **1A**) and $\delta_{\rm H}$ 5.19 (d, J = 1.3, 11.7 Hz, H-6', **1B**) show long-range correlations with the carbon signals of C-4 ($\delta_{\rm C}$ 150.3 (**1A**), $\delta_{\rm C}$ 147.4 (**1B**)) and C-11 ($\delta_{\rm C}$ 36.9 (**1A**), $\delta_{\rm C}$ 38.4 (**1B**)), respectively. The downfield shifts of position 6 ($\delta_{\rm H}$ 4.82, **1A**; $\delta_{\rm H}$ 5.19, **1B**) revealed the presence of a hydroxyl group at C-6. When compared with previously isolated sesquiterpenoids from this plant, these data indicated that **1** is very similar to linderene or hydroxyl-



1A; R₁: H, R₂:α-OH; 2A R₁:COCH₃, R₂: α-OH
1B; R₁:H, R₂:β-OH; 2B R₁:COCH₃, R₂: β-OH
Figure 1. Structure of compounds 1A, 1B, 2A, and 2B.



Figure 2. Stereostructure and NOE correlation of compound(s) 1A and 1B generated by the CAChe system.

indestenolide,⁶ which were previously isolated from *L. strychnifolia*.

The H–H coupling constants between H-5 and H-6 were J = 10.5 Hz (**1A**) and J = 11.7 Hz (**1B**), respectively, showing the *trans* axial–axial relationship of these protons. The H-6 proton signal showed a cross-peak with H-14 ($\delta_{\rm H}$ 0.61, **1A**; $\delta_{\rm H}$ 1.23, **1B**) in the NOESY spectrum (Figure 2), indicating the same orientation of H-6 and the C-14 methyl group. Furthermore, as can be seen in Figure 2, NOE

^{*} To whom correspondence should be addressed. Tel and Fax: +81 95 848 4387. E-mail: ikouno@net.nagasaki-u.ac.jp.

Table 1. ¹³C and ¹H NMR Spectral Data of Compounds 1A and 1B (C₅D₅N) and 2A and 2B (CDCl₃) (¹³C, 125 MHz; ¹H, 500 MHz)

	1A		2A		1B		2B	
position	¹³ C	¹ H	¹³ C	¹ H	¹³ C	¹ H	¹³ C	¹ H
1	28.8	1.33 (dt, J=3.5, 7.5 Hz)	29.2	1.41 (dt, J= 3.5, 7.5 Hz)	27.4	1.30 (dt, J = 3.5, 7.5 Hz)	27.9	1.31 (dt, $J = 3.5$, 7.5 Hz)
2a 2b	15.9	0.78 (m)	16.8	0.83 (m) 0.77 (dt, J=3.5, 5.4 Hz)	15.7	1.08 (dt, J = 3.5, 7.5 Hz) 0.79 (m)	16.7	0.82 (m) 0.77 (dt, J=3.5, 5.4 Hz)
3 4	23.1 150.3	2.01 (m)	23.6 148.6	2.03 (m)	22.7 148.6	2.01 (m)	23.5 147.4	1.96 (m)
5	61.9	4.12 (d, $J = 10.5$ Hz)	59.6	3.68 (t, J = 2.5, 10.9 Hz)	69.9	3.00 (dt, <i>J</i> = 2.0, 11.7 Hz)	67.2	2.86 (d, $J = 12.4$ Hz)
6	61.7	4.82 (d, $J = 10.5$ Hz)	63.4	5.70 (d, <i>J</i> = 10.9 Hz)	67.3	5.19 (dd, <i>J</i> = 1.3, 11.7 Hz)	69.7	5.61 (d, $J = 12.4$ Hz)
7	159.7		154.0		163.3		157.6	
8	105.7		105.4		105.7		105.9	
9a 9b	49.9	2.43 (d, $J = 14.0$ Hz) 2.83 (d, $J = 14.0$ Hz)	49.9	2.56 (d, $J = 14.2$ Hz) 2.28 (d, $J = 14.2$ Hz)	47.3	2.74 (d, $J = 13.0$ Hz) 1.97 (d, $J = 13.0$ Hz)	47.5	2.60 (d, J = 13.9 Hz) 1.85 (d, J = 13.9 Hz)
10	36.9		37.7		37.2		38.4	
11	125.5		130.6		121.2		122.5	
12	171.7		170.7		171.9		171.1	
13	7.3	1.94 (s)	9.2	1.94 (s)	8.4	2.38 (d, $J = 1.3$ Hz)	8.3	1.84 (s)
14	20.7	0.61 (s)	21.5	0.58 (s)	17.4	1.23 (d, $J = 1.8$ Hz)	17.9	0.99 (s)
15a	106.7	5.66 (d, $J = 1.8$ Hz)	107.5	5.04 (br s)	108.4	5.81 (d, $J = 1.8$ Hz)	108.9	5.07 (d, $J = 1.6$ Hz)
15b		5.27 (d, $J = 1.8$ Hz)		4.77 (br s)		5.28 (d, $J = 1.8$ Hz)		4.80 (d, $J = 1.6$ Hz)
-CO			171.9					171.9
-OAc			21.0	2.10 (s)			20.8	2.20 (s)

correlations between H-9a and H-14, H-9b and H-5, and H-1 and H-9b were observed, suggesting the cyclopropane ring is in the same orientation as the C-14 methyl group. From these data, it was concluded that, in C_5D_5N solution, **1** is a mixture of the 8α -OH and 8β -OH epimers at the C-8 hemiketal position of eudesmanolide.

A NOE correlation was found between H-6 and the H₃-13 methyl signals only in the case of **1A** in the NOESY of **1**. The stereostructure generated by the CAChe system⁷ and the Dreiding model showed a proximity between H-6 and the H₃-13 methyl group only in **1A**, as shown in Figure 2. In addition, the H₃-14 methyl signal at $\delta_{\rm H}$ 1.23 of **1B** was considerably deshielded relative to that of **1A** at $\delta_{\rm H}$ 0.61 due to the anisotropic effect of the 8 β -OH oxygen atom, supporting the *anti*-relationship of the C-14 methyl group and the 8-OH in the case of **1A** and the *syn*-structure for **1B**.

On the basis of these data, it was concluded that compound **1** in CDCl₃ should be the *anti*-structure with respect to the C-14 methyl group and the 8-OH, since the chemical shift of H₃-14 at $\delta_{\rm H}$ 0.52 was more similar to that of **1A** ($\delta_{\rm H}$ 0.61) than that of **1B** ($\delta_{\rm H}$ 1.23). Compound **1** was given the trivial name strychnistenolide. **1A** was named strychnistenolide B.

Compound 2 was obtained as colorless powder. EIMS showed a molecular ion peak at m/z 304 [M⁺], 42 mu more than that of 1. The ¹H and ¹³C NMR spectra (Table 1) in CDCl₃ strongly resembled those of **1A** and **1B** in C₅D₅N, except for the acetyl methyl signals at $\delta_{\rm H}$ 2.10 (3H, s) (2A) and $\delta_{\rm H}$ 2.20 (3H, s) (2B). The position of the acetate at C-6 was supported by the downfield shifts of the H-6 proton signal at $\delta_{\rm H}$ 5.70 (2A) and $\delta_{\rm H}$ 5.61 (2B) in the ¹H NMR spectrum of 2. Thus, 2 was determined to be the 8-OH epimeric mixture of 6-O-acetyl strychnistenolide, as shown in Figure 1. All signals in the ¹H and ¹³C NMR spectra of a mixture of 2 (2A and 2B) were assigned by analysis of the individual ¹H-¹H COSY, HSQC, and HMBC spectra. The assignments are shown in Table 1. As was the case of 1, the most downfield H₃-14 methyl signal at $\delta_{\rm H}$ 0.99 was assigned to ${f 2B}$ and the H₃-14 methyl signal at $\delta_{
m H}$ 0.58 to 2A. The strong similarity of the ¹H and ¹³C NMR spectra



Figure 3. Structure of compound 3 and NOE correlations.

for **2** and **1** led us to determine the stereochemistry of **2A** and **2B** as shown in Figure 1.

Compound **3** was obtained as colorless rods, mp 181– 182 °C. EIMS showed a molecular ion peak at m/z 324 [M⁺], and the ¹³C NMR spectrum showed 17 carbon signals. These data, together with elemental analysis, established the molecular formula to be $C_{17}H_{24}O_6$. The ¹H NMR spectrum of **3** shows three CH₃ signals at δ_H 1.31, 1.66, and 2.02 and signals for an ethoxy group at δ_H 1.26 (3H), 4.22 (1H), and 4.17 (1H). The ¹H-¹H COSY confirmed the presence of an oxygenated methylene group (H₂-9) and a cyclopropane ring (H-1, -2, and -3). The presence of an ester and lactone groups was supported by ¹³C NMR signals at δ_C 167.1 and 169.5 and the absorptions at ν_{max} 1751 and 1641 cm⁻¹ in the IR spectrum.

Extensive analyses of ¹H⁻¹H COSY, HSQC, and HMBC spectra of **3** established the structure shown in Figure 3. In the HMBC spectrum, no connectivity was observed between the H-9 methylene group and the C-8 carbonyl carbon, indicating that C-4 and C-8 are joined by a lactone ring. This was confirmed by the proton chemical shifts of the H-9 methylene group at $\delta_{\rm H}$ 3.51 and 3.48. The NOE correlations in the NOESY spectrum of **3**, H-1 to H₂-9, H-6 to H₃-14, H₃-13 to H-6, H₂-9 to H3-15, and H₃-15 to H-3 as shown in Figure 3, confirmed the stereostructure, except for the configuration at C-4 and C-6.



Figure 4. ORTEP drawing of compound 3.

The structure of **3** was established by a single-crystal X-ray analysis, which also revealed the configuration of C-4. A single crystal suitable for X-ray analysis with a space group $P2_12_12_1$ (Z = 4) was obtained by recrystallization from CHCl₃. After the data collection was performed on a RIGAKU AFC 7S diffractometer, the structure was solved by direct methods (SIR92),⁸ then refined by full-matrix least-squares based 1952 observed reflections ($I > 3.00\sigma[I]$). Hydrogen atoms were refined isotropically, and nonhydrogen atoms were treated anisotropically. The final refinement converged at R = 0.032. This led to the suggested structure (Figure 3) including the configuration of 6β -OH. The ORTEP drawing of **3**, named strychnilactone, is shown in Figure 4.

More than 10 secoeudesmane-type sesquiterpenes have been identified so far, but **3** is unique and represents the first example of a secoeudesmanolide in which the C-8, C-9 bond has been cleaved with rearrangement to form a C-8, C-4 δ -lactone.

Experimental Section

General Experimental Procedures. Optical rotation was recorded with a JASCO DIP-370 polarimeter. ¹H and ¹³C NMR spectra were measured on a Varian Gemini 300 spectrometer operating at 300 MHz for ¹H and 75 MHz for ¹³C for compound 1 in CDCl₃ and a Varian UNITY plus 500 spectrometer for other compounds, operating at 500 MHz for ¹H and 125 MHz for ¹³C, including NOESY, and ¹H-¹H COSY, HSQC, and HMBC with field gradient technique in CDCl₃ or C₅D₅N. Chemical shifts were given on a δ (ppm) scale with TMS as an internal standard. EIMS were measured on a JEOL JMS-DMX-303 mass spectrometer. Column chromatography was carried out by using silica gel 60 (Merck, Art. 7734 and Art. 9385). MPLC was performed on Kusano C.I.G. prepacked Si-5 column (Kusano Kagakukikai Co., 22 mm i.d. × 100 mm).

Plant Material. The roots of *L. strychnifolia* were collected in the autumn of 1995 at the Botanical Garden, Nagasaki University, and identified by Dr. T. Ikenaga (former Assoc. Prof. of Botanical Garden). A voucher specimen (identification number NAP 20-06-95/2) was deposited at the Laboratory of Botanical Garden belonging to the Faculty of Pharmaceutical Sciences, Nagasaki University.

Extraction and Isolation. The roots (2.9 kg) of *L. strychnifolia* were extracted three times with Et₂O, followed by three times with MeOH. The Et₂O extract was evaporated in vacuo to give a residue (53.7 g). The Et₂O-soluble (33.3 g) and EtOAc-soluble (0.37 g) portions of the MeOH extracts were combined with the Et₂O extract. The combined residual part was suspended in C₆H₆ to give a supernatant, which was evaporated in vacuo to give a residue (42.2 g). This residue was chromatographed on silica gel, and the fractions eluted with n-hexanes—EtOAc (19:1) and MPLC with CHCl₃–n-hexane (3: 2) successively to afford compounds **1** (416.6 mg), **2** (100.5 mg), and **3** (17.1 mg), along with known sesquiterpenes that were

identified by comparison of their spectroscopic data with those in the literature.

Strychnistenolide (1): colorless needles (*n*-hexanes– EtOAc), mp 185–186 °C; $[\alpha]_D^{20}$ +36.3° (*c* 0.3, MeOH); IR (CHCl₃) ν_{max} 3394, 1752, 1600 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 0.52 (s, H-14), 0.76 (m, H-2a), 0.83 (m, H-2b), 1.40 (m, H-1), 1.92 (s, H-13), 2.05 (m, H-3), 2.21 (d, *J* = 14.0 Hz, H-9a), 2.50 (d, *J* = 14.0 Hz, H-9b), 3.41 (d, *J* = 10.6 Hz, H-5), 4.42 (d, *J* = 10.6 Hz, H-6), 5.11 (br s, H-15a), and 5.14 (br s, H-15b); ¹³C NMR (CDCl₃, 75 MHz) δ 8.5 (C-13), 16.8 (C-2), 21.6 (C-14), 23.8 (C-3), 29.2 (C-1), 37.8 (C-10), 49.5 (C-9), 62.7 (C-6), 62.8 (C-5), 105.9 (C-8), 107.7 (C-15), 128.5 (C-11), 149.6 (C-4), 157.9 (C-7), 172.8 (C-12); EIMS *m*/*z* 262 [M⁺], 244 [M–H₂O]⁺; *anal.* C 68.82%, H 6.86%, calcd for C₁₅H₁₈O₄, C 68.68%, H 6.92%.

Strychnistenolide 6-*O***-acetate (2):** colorless amorphous powder (*n*-hexanes–EtOAc); ¹H and ¹³C NMR (CDCl₃, 500 MHz), (see Table 1); EIMS m/z 304 [M]⁺, 286 [M–H₂O]⁺.

Strychnilactone (3): colorless rods (CHCl₃–MeOH); mp 181–182 °C; $[\alpha]_D^{20}$ –267.6° (*c* 0.2, MeOH); IR (KBr) ν_{max} 3357, 1751, 1641 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 0.51 (dt, *J* = 5.5, 8.4 Hz, H-2a), 1.25 (m, H-1), 1.26 (t, *J* = 7.1 Hz, -COOCH₂CH₃), 1.31 (s, H-14), 1.49 (ddd, *J*=3.7, 6.1, 8.4 Hz, H-3), 1.58 (m, H-2b), 1.66 (s, H-15), 2.02 (s, H-13), 2.10 (d, H-5), 3.48 (d, *J* = 10.3 Hz, H-9a), 3.51 (d, *J* = 10.3 Hz, H-9b), 4.17 (dq, *J* = 7.1, 10.8 Hz, -COOCH₂CH₃), 4.22 (dq, *J* = 7.1, 10.8 Hz, -COOCH₂CH₃), 4.68 (d, *J* = 4.1 Hz, H-6); ¹³C NMR (CDCl₃, 125 MHz) δ 5.1 (C-2), 13.6 (-COOCH₂CH₃), 15.9 (C-13), 18.3 (C-14), 27.1 (C-3), 27.2 (C-1), 32.9 (C-15), 49.3 (C-10), 53.6 (C-5), 61.8 (-COOCH₂CH₃), 65.4 (C-6), 73.4 (C-9), 92.7 (C-4), 133.4 (C-11), 134.8 (C-7), 167.1 (C-8), 169.5 (C-12); EIMS *m*/z 324 [M]⁺; *anal.* C 63.29%, H 7.29%, calcd for C₁₇H₂₄O₆, C 62.95%, H 7.46%.

X-ray Crystal Structure Analysis of Strychnilactone (3). Crystal data: $C_{17}H_{24}O_6$, MW = 324.37; orthorhombic space group $P2_12_12_1$; a = 14.912(2) Å, b = 15.524(2) Å, c = 7.368(2)Å, V = 1705.5(6) Å³, Z = 4, $D_{calc} = 1.263$ g/cm³, Mo K α ($\lambda =$ 0.71069 Å). The reflection data were collected on a Rigaku RASA 7S autodiffractometer using graphite-monochromated Mo K α radiation with the $\omega - 2\theta$ scan technique to a maximum θ of 70° at room temperature (23 °C), independent reflections 2275, observed number of reflections 1925 [$F > 3.0\sigma(F)$]. The structure was solved by the direct method using SIR92. All atomic parameters, with anisotropic temperature factors for non-hydrogen atoms and isotropic ones for hydrogen atoms, were refined by the full-matrix least-squares method.

Acknowledgment. We express our appreciation to Mr. K. Inada, Mr. N. Yamaguchi, and Mr. M. Ohwatari of the Center for Instrumental Analysis of Nagasaki University for performing NMR, EIMS, and elemental analysis experiments, respectively, and Ms. J. Nagaoka of Nagasaki University for X-ray measurements.

References and Notes

- Ishii, H.; Tozyo, T.; Nakamura, M.; Takeda, K. Tetrahedron 1968, 24, 625-631.
- (2) Takeda, K.; Ishii, H.; Tozyo, T.; Minato, H. J. Chem. Soc. (C) 1969, 1920–1921.
- (3) Takeda, K.; Horibe, I.; Teraoka, M.; Minato, H. J. Chem. Soc. (C) 1969, 2786–2788, and references therein.
- (4) Kouno, I.; Hirai, A.; Jiang, Z.-H.; Tanaka, T. Phytochemistry 1997, 46, 1283–1284.
- (5) Kouno, I.; Hirai, A.; Fukushige, A.; Jiang, Z.-H.; Tanaka, T. Chem. Pharm. Bull. 1999, 47, 1056–1057.
- (6) Takeda, K.; Horibe, I.; Minato, H. J. Chem. Soc. (C) 1968, 569-572.
- (7) CAChe 3.2 for Windows 95/98/NT 4.0, Oxford Molecular Ltd., 1999.
- (8) Altomare, A.; Burla, M. C.; Camalli, M.; Cascarano, M.; Giacovazzo, C.; Guagliardi, A.; Polidori, G. J. Appl. Crystallogr. 1994, 27, 435.

NP000154S